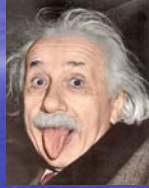




### History

- 1917 '*Emission by Stimulation*' Einstein
- 1954 MASER
- 1960 'Red Ruby Laser' Theodore Maiman
- 1965 Ruby Laser used in the inner ear of Pigeons
- 1967 Nd:YAG used on otosclerotic foot plate
- 1972 Argon laser used on otic capsule
- 1980 CO<sub>2</sub> Laser



## LASER

- Light
- Amplification
- Stimulated
- Emission
- Radiation

## Light

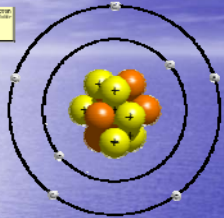
**Electromagnetic Spectrum Diagram**

Wavelength (nanometers)

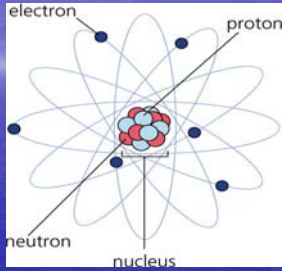
The diagram shows the electromagnetic spectrum with wavelength in nanometers on a logarithmic scale from 10<sup>4</sup> to 10<sup>12</sup>. The visible light spectrum is highlighted in a rainbow, ranging from 400 nm (violet) to 700 nm (red).

Three images illustrating light phenomena: a rainbow, a purple laser beam, and a colorful fiber optic display.

## Atom

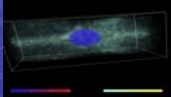


• Particle




electron  
proton  
neutron  
nucleus

• Cloud




Uncertainty Principle




Heisenberg, Werner

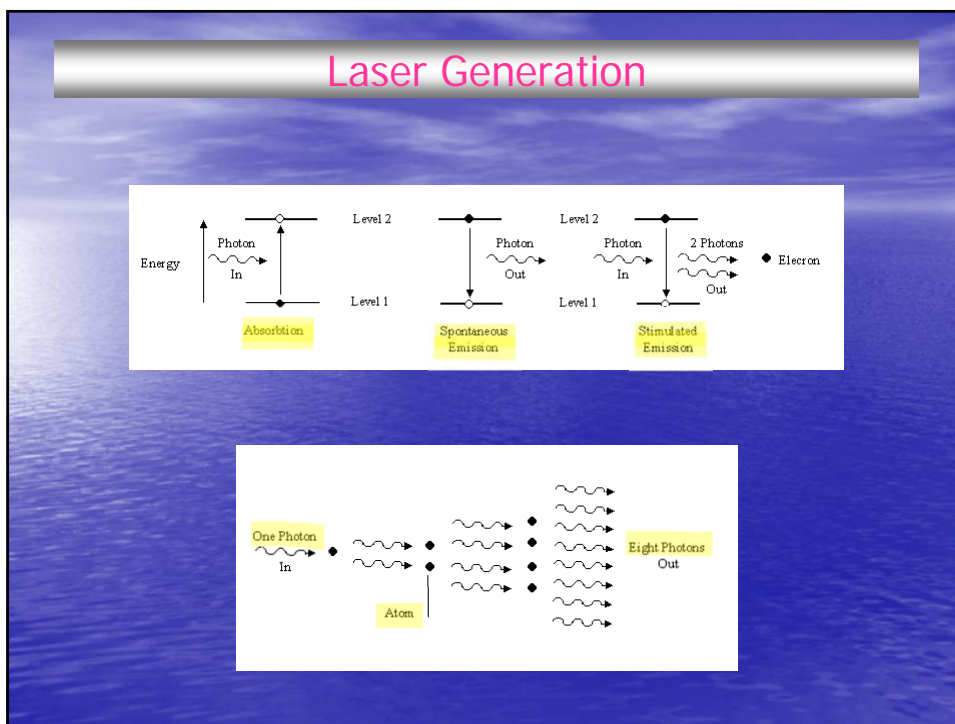
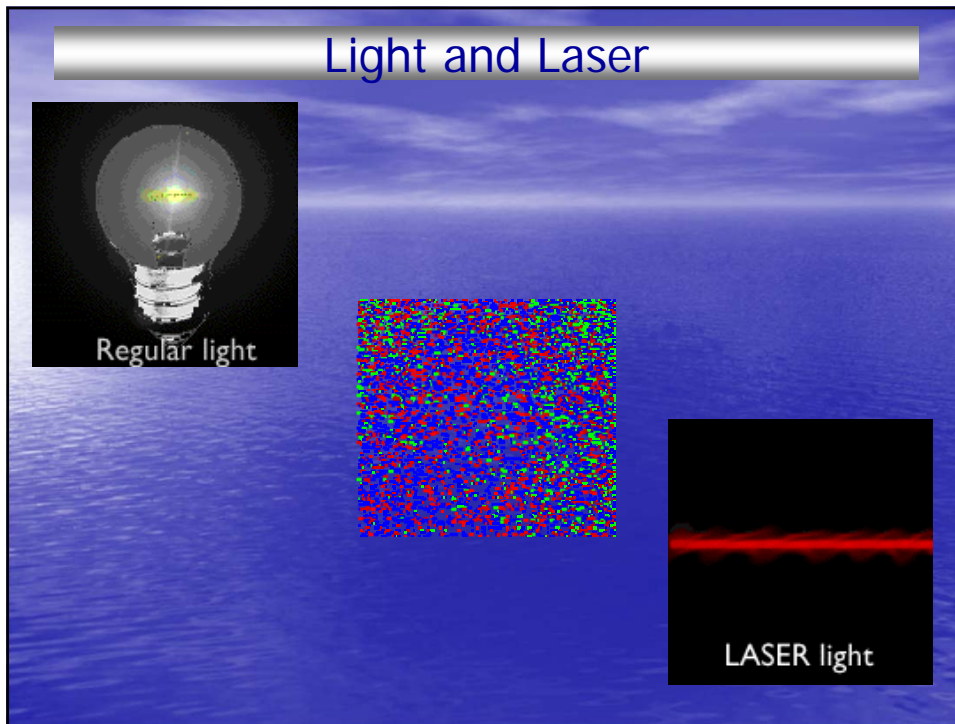
• String



## Generation of Photon

A quantum of electromagnetic energy having both particle and wave properties. A photon has no electric charge or mass but possesses momentum, energy, and spin





### Laser generation

The diagram illustrates the stages of laser generation through three energy level diagrams and a wave representation:

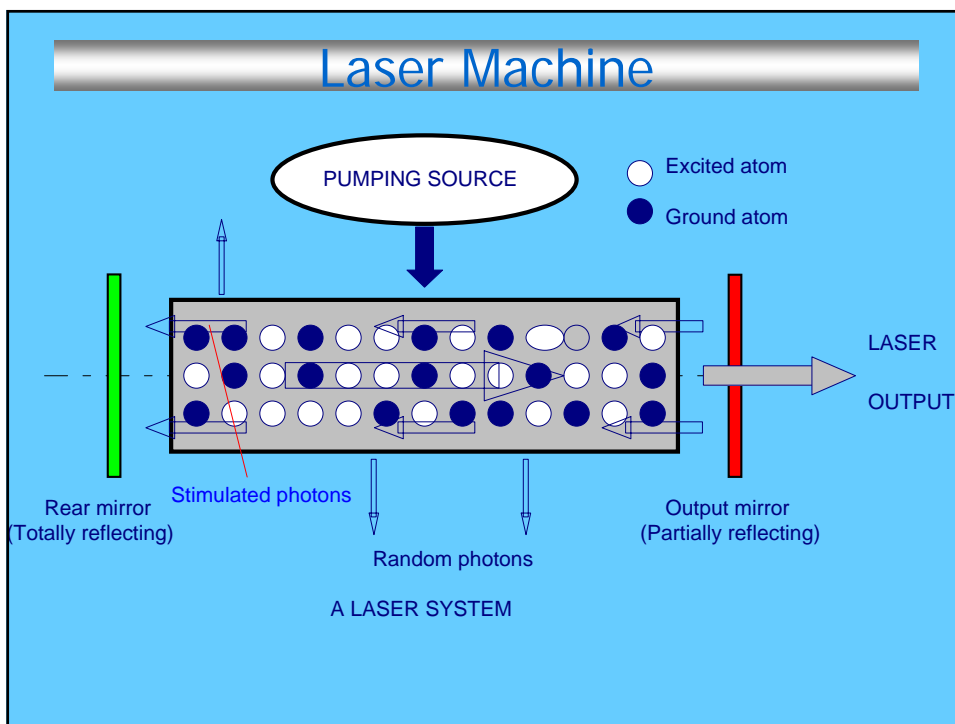
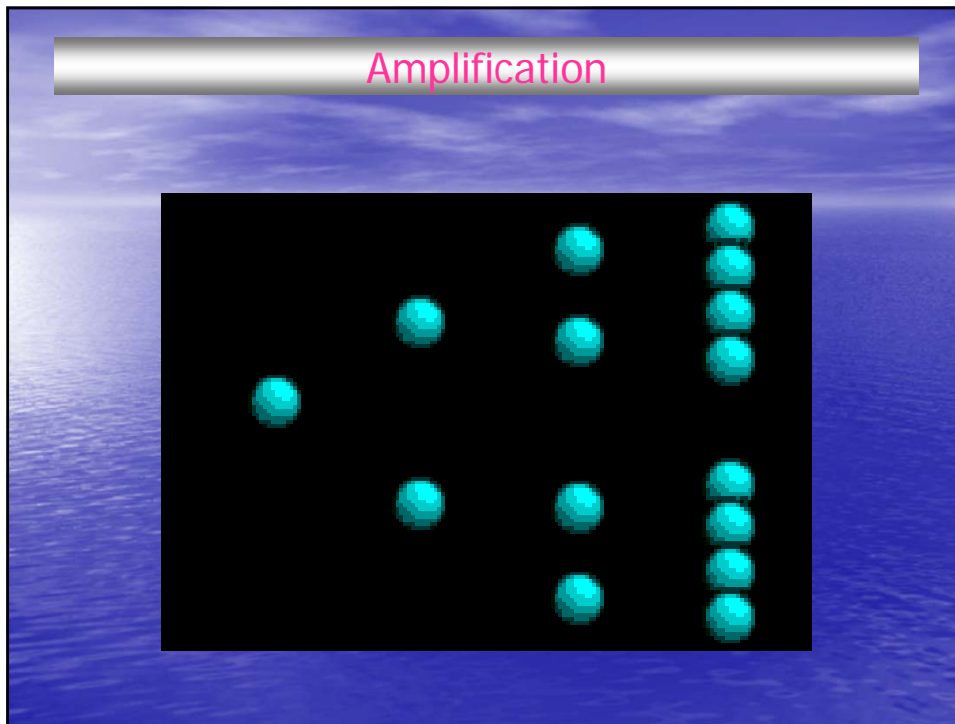
- Non inverted population:** Shows a ground state with five electrons (e) and an excited state with two electrons. Transitions are shown as downward arrows.
- Inverted population:** Shows the ground state with two electrons and the excited state with five electrons. This state is necessary for laser action.
- spontaneous emission:** Shows an electron in the excited state transitioning to the ground state, emitting a photon.

A 3D wave representation shows the electric field (E) and magnetic field (H) components of the emitted light.

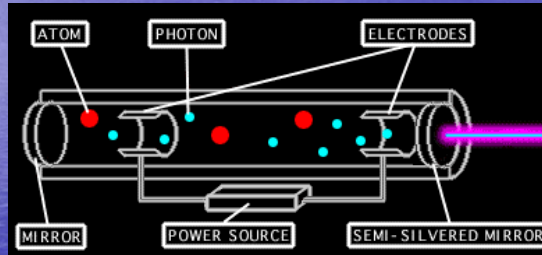
### Stimulated Emission

The diagram illustrates the process of stimulated emission:

- An **ATOM** (red sphere) is shown in an excited state.
- An incoming **PHOTON** (yellow arrow) stimulates the atom to transition to a lower energy state.
- The atom emits a second photon (yellow arrow) that is identical in phase and direction to the incoming photon.

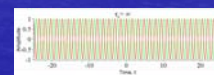
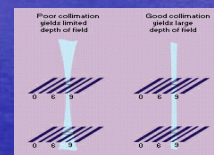


## Laser Machine



## Properties of Laser

- **Monochromatic** : EM radiation of same wavelength
- **Collimated**: Parallel waveform
- **Coherence**: Waves in phase; both time and Space
- **Beam size**



## Laser Energy and Irradiance

**Power** (Watts) = **Energy** (Joules) / **Time** (Sec)

**Power Density**

**Irradiance** = Power / Unit area

**I** = Watts / cm<sup>2</sup>



## Laser Classifications

- **Class 1** Inherently safe
- **Class 2** Low power, within the visible band. Safety by aversion response
- **Class 3A** Hazardous to eye if viewed through microscope
- **Class 3B** Direct viewing is hazardous. Viewing reflection is safe
- **Class 4** High powered hazardous laser



## General Concepts

- **All lasers consist of**
  - A medium with suitable set of energy levels
  - A pumping (energy) source
  - An optical cavity
- **Classified by active medium**
  - Solid state lasers
  - Gas lasers
  - Liquid lasers
  - Semiconductor lasers
- **Compared by output characteristics**
  - Wavelength
  - Coherence
  - Beam divergence
  - Irradiance or power

## Types of Lasers

- Gas
- Liquid
- Solid
- Semiconductors

## Types of Lasers

- CO<sub>2</sub> Laser
- Nd:YAG Laser
- KTP laser
- Argon laser
- Holmium : YAG laser
- Erbium : YAG laser
- Diode laser
- Ruby laser

## Gas Laser

### Gas Lasers

Nitrogen (UV) 337 nm Spectroscopy, Non-destructive Testing  
 Helium Cadmium (UV) 325 nm Spectroscopy, Non-destructive Testing  
     (Violet) 441 nm Spectroscopy, Non-destructive Testing  
**Argon** (Blue) 488 nm Displays, Lightshows, Ophthalmology, Surgery  
     (Green) 514 nm Displays, Lightshows, Ophthalmology, Surgery  
**Krypton** (Blue) 476 nm Displays, Lightshows  
     (Green) 528 nm Displays, Lightshows  
     (Yellow) 568 nm Displays, Lightshows  
     (Red) 647 nm Displays, Lightshows, Ophthalmology  
 Xenon (White) multiple Invasive Surgery  
 Helium Neon (Green) 543 nm Industrial Alignment, Dynamic Balancing  
     (Yellow) 594 nm Industrial Alignment  
     (Orange) 612 nm Industrial Alignment  
     (Red) 633 nm Industrial Alignment, Dynamic Balancing, Printing, Copiers  
     (NIR) 1,152 nm Spectroscopy  
     (MIR) 3,390 nm Spectroscopy  
 Hydrogen Fluoride (MIR) 2,700 nm Non-Destructive Testing  
**Carbon Dioxide** (FIR) 10,600 nm Industrial Cutting, Soldering/Welding, Materials Processing, surgery

## Solid State Lasers

### Solid State Lasers

Doubled Nd:YAG (Green) 532 nm Invasive Surgery, Light Shows, Military Ranging

Neodymium:YAG (NIR) 1,064 nm Industrial Sealing, Soldering, Lithography, Invasive Surgery

Erbium:Glass (MIR) 1,540 nm Spectroscopy

Erbium:YAG (MIR) 2,940 nm Invasive Surgery, Dentistry

Holmium:YLF (MIR) 2,060 nm Spectroscopy

Holmium:YAG (MIR) 2,100 nm Invasive Surgery, Dentistry

Chromium Sapphire (Ruby) (Red) 694 nm Invasive Surgery, Cosmetic Surgery

Titanium Sapphire (NIR) 840-1,100 nm Spectroscopy

Alexandrite (NIR) 700-815 nm Spectroscopy

## Excimers

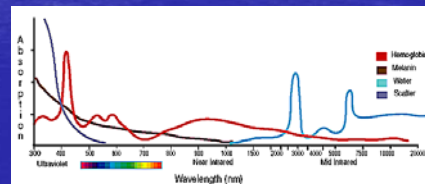
### Excimer Gas Lasers

- Argon Fluoride (UV) 193 nm Ophthalmology, Industrial Cutting/Drilling, Krypton Chloride (UV) 222 nm
- Krypton Fluoride (UV) 248 nm
- Xenon Chloride (UV) 308 nm Invasive Surgery, Industrial Cutting
- Xenon Fluoride (UV) 351 nm

## Wave length and Absorption

Laser Type	Wavelength (nm)	Spectral Region	Mode	Typical Maximum Power
CO <sub>2</sub>	10,600	Mid infrared	CW and gated	100 W cw
Holmium	2,100	Near infrared	Pulsed	15 W avg.
Nd:YAG	1,064	Near infrared	CW and pulsed	100 W cw
Diode	800-890	Near infrared	CW	>4 W
KTP/KDP	532	Visible	Pulsed	20 W avg.
Argon	488/514	Visible	CW	20 W
Excimer ArF-XeCL	190	Ultraviolet	Pulsed	550 mJ
	308	Ultraviolet	Pulsed	250 mJ

CW = continuous wave.



## Laser Tissue Interaction

- Laser absorption depends on the absorption coefficient of the tissue
- Photothermal Effect: Laser energy is converted into heat which in turn raises the temperature of the Cytosol and Extracellular fluid (water)
- 49°C **Photocoagulation**
- 55° C **Irreversible Tissue damage**
- 100° C **Vaporisation**
- 350° C **Charring**

## Ruby Laser

First working laser to be demonstrated

Cylindrical crystal of synthetic **Sapphire**  
( $\text{Al}_2\text{O}_3$ )

doped with roughly 0.05%, by weight, of  
**Chromium ions ( $\text{Cr}^{3+}$ ) = Ruby**

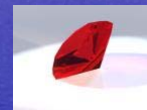
## Ruby Laser

**Lasing medium** Ruby Crystal

**Wave Length** 694 nm

**Power / Energy** 3 J

**Beam Delivery** Fibre Optic

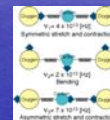


## Ruby Laser

- Ruby absorbs pumping energy in **blue-green** region of the spectrum and emits laser energy at a **wavelength of 694.3 nm**
- The absorption of blue-green light raises the  $\text{Cr}^{3+}$  ions to a broad band of upper levels
- A population inversion is established between this state and the ground state

## Carbon Dioxide Laser

- The important energy levels are provided not by the distribution of electrons
  - Instead from the vibration of the entire  $\text{CO}_2$  molecule itself
- The  $\text{CO}_2$  molecule can be pictured as
  - a linear arrangement of O-C-O atoms,
  - which vibrate in relation to each other
- Several different modes of vibration give rise to a set of energy levels with transitions far into the infra-red.
- The principal  $\text{CO}_2$  wavelength is **10.6  $\mu\text{m}$** 
  - Far Infra-red



## Carbon Dioxide Laser

- **Lasing medium**     $\text{CO}_2$
- **Wave Length**     $10.6 \mu\text{m}$
- **Power / Energy**    10-60 W
- **Superpulse**        250 W
- **Beam Delivery**    Articulated arm

## Argon-ion Laser

- Lasing takes place in the plasma state
- An electric discharge is created in a narrow tube of gaseous argon.
- Argon atoms are first ionised and then excited by multiple collisions with electrons into their upper energy levels.

## Argon-ion Laser

Lasing medium	Argon gas
Wave Length	488-514 nm
Power / Energy	3-10 W
Beam Delivery	Fibre optic
Light	Blue 488nm    Green 514

## Nd-YAG Laser

Lasing medium **Yttrium-Aluminium-Garnate**  
( $\text{Y}_3\text{Al}_5\text{O}_{12}$ )

Doped with 0.7% by weight of Neodymium ( $\text{Nd}^{3+}$ ) ions



Wave Length	1064 nm
Power / Energy	100 W
Beam Delivery	Fibre optic

Note: Q switched Nd-YAG Laser



## KTP laser

<b>Lasing medium</b>	Nd:YAG frequency doubled by passing it through <b>Potassium Titalyl Phosphate (KTiOPO<sub>4</sub>)</b>
<b>Wave Length</b>	532 nm
<b>Power / Energy</b>	15 W
<b>Beam Delivery</b>	Fibre optic



## Holmium:YAG laser

<b>Wavelength:</b>	2.1 micron
<b>Pulse duration:</b>	250 ms
<b>Peak power:</b>	11 KW
<b>Aiming beam:</b>	5mW HeNe
<b>Delivery system:</b>	Fibreoptic



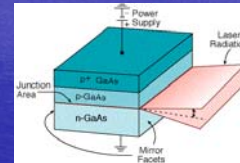
## Diode Laser

**Lasing medium** Gallium-Arsenide / Semiconductors

**Wave Length** 810nm

**Power / Energy** > 4 W

**Beam Delivery** Fibre optic



## Erbium Laser

- Lasing Medium: Erbium :YAG

- Wave Length: 10.6

- Power / Energy: 0.5-1.0J

- Beam Delivery: Fibre optic



## Recommended Reading

<http://www.ncbi.nlm.nih.gov/pmc/articles/PMC3799025/pdf/islsm-20-095.pdf>

<http://www.sciencedirect.com/science/article/pii/S0001209212002578>